



National Student Team Contest (first stage)
Solution of task 8. Colored nanofilms

The color of the film is governed by interference maximum condition:

$$\lambda = 2Dn \tag{1}$$

Since thickness of the film wasn't changed, we get an expression by dividing eq. (1) written for both cases:

$$\frac{\lambda_{\text{green}}}{\lambda_{\text{violet}}} = \frac{n_{\text{green}}}{n_{\text{violet}}} = \frac{530}{430} = 1.23 \tag{2}$$

Then we need to use Bruggeman effective media approximation. According to it one can obtain for two component system:

$$(1 - P) \frac{\epsilon_S - \epsilon_{\text{eff}}}{\epsilon_S + 2\epsilon_{\text{eff}}} + P \frac{1 - \epsilon_{\text{eff}}}{1 + 2\epsilon_{\text{eff}}} = 0 \tag{3}$$

$$(P - 1)(\epsilon_S - \epsilon_{\text{eff}})(1 + 2\epsilon_{\text{eff}}) = P(1 - \epsilon_{\text{eff}})(\epsilon_S + 2\epsilon_{\text{eff}}) \tag{4}$$

$$(P - 1)(\epsilon_S - \epsilon_{\text{eff}} + 2\epsilon_S \epsilon_{\text{eff}} - 2\epsilon_{\text{eff}}^2) = P(\epsilon_S - \epsilon_{\text{eff}} \epsilon_S + 2\epsilon_{\text{eff}} - 2\epsilon_{\text{eff}}^2) \tag{5}$$

$$P(\epsilon_S - \epsilon_{\text{eff}} + 2\epsilon_S \epsilon_{\text{eff}} - 2\epsilon_{\text{eff}}^2) - (\epsilon_S - \epsilon_{\text{eff}} + 2\epsilon_S \epsilon_{\text{eff}} - 2\epsilon_{\text{eff}}^2) = P(\epsilon_S - \epsilon_{\text{eff}} \epsilon_S + 2\epsilon_{\text{eff}} - 2\epsilon_{\text{eff}}^2) \tag{6}$$

$$\epsilon_S - \epsilon_{\text{eff}} + 2\epsilon_S \epsilon_{\text{eff}} - 2\epsilon_{\text{eff}}^2 = P(3\epsilon_{\text{eff}} \epsilon_S - 3\epsilon_{\text{eff}}) \tag{7}$$

$$2\epsilon_{\text{eff}}^2 + \epsilon_{\text{eff}}(3P(\epsilon_S - 1) + 1 - 2\epsilon_S) - \epsilon_S = 0 \tag{8}$$

or

$$P = \frac{\epsilon_S - \epsilon_{\text{eff}} + 2\epsilon_S \epsilon_{\text{eff}} - 2\epsilon_{\text{eff}}^2}{3\epsilon_{\text{eff}}(\epsilon_S - 1)} \tag{9}$$

One can obtain for first case of green film:

$$2\epsilon_{\text{eff}}^2 + \epsilon_{\text{eff}}(0.6(3 \cdot 2.1) + 1 - 2 \cdot 3.1) - 3.1 = 0 \tag{10}$$

or

$$2\epsilon_{\text{eff}}^2 - 1.42 \cdot \epsilon_{\text{eff}} - 3.1 = 0 \tag{11}$$

then

$$\epsilon_{\text{eff}} = \frac{1.42 \pm \sqrt{1.42^2 + 4 \cdot 2 \cdot 3.1}}{4} = 1.65 \tag{12}$$

then

$$n_{\text{green}} = \sqrt{\epsilon_{\text{eff}}} = 1.28 \quad (13)$$

Let's move to violet film.

$$n_{\text{violet}} = \frac{n_{\text{green}}}{1.23} = \frac{1.28}{1.23} = 1.04 \quad (14)$$

$$\epsilon_{\text{vio}} = n_{\text{violet}}^2 = 1.08 \quad (15)$$

$$P = \frac{3 \cdot 1 - 1.08 + 2 \cdot 3 \cdot 1 \cdot 1.08 - 2 \cdot 1.08^2}{3 \cdot 1 \cdot 0.8(3 \cdot 1 - 1)} = \frac{2.02 + 6.7 - 2.33}{6.8} = 0.94 \quad (16)$$

Therefore porosity of new film is equal to 94%.